



NATIONAL LEVEL SCIENCE TALENT SEARCH EXAMINATION - 2015 (UPDATED)

Paper Code: UN412 Solutions for Class: 11 (PCM)



 \Rightarrow Minimum number of elements in A \cup B is 6. We have, $7^{103} = 7(49)^{51} = 7(50 - 1)^{51}$ **(C)** = $7(50^{51} - {}^{51}C_1 50^{50} + {}^{51}C_2 50^{49} - \dots - 1)$ $= 7(50^{51} - {}^{51}C_1 50^{50} + {}^{51}C_2 50^{49} - ...) - 25 + 18$ = k + 18 (say), since k is divisible by 25, \therefore Remainder is 18. 5. **(A)** Let f(x) = x and $g(x) = x^3$. Then f(x) g(x) $= x^4$, even; option (A) is false. Let f(x) and g(x) are two even functions and $\mathbf{F}(x) = \mathbf{f}(x) + \mathbf{g}(x)$ F(-x) = f(-x) + g(-x) = f(x) + g(x) = F(x) \Rightarrow F(*x*) is even, option (B) is true. Let f(x) be an even function and g(x) be an odd function and F(x) = f(x) g(x)F(-x) = f(-x) g(-x) = f(x) (-g(x)) \Rightarrow -F(x) is odd, option (C) is true (D) $g(-x) = \frac{f(-x) + f(x)}{2} = \frac{f(x) + f(-x)}{2}$ = g(x) is even, option (D) is ture. $T_{_{20}}$ = (20^{th} term of 2, 4, 6, 8, ...) \times (20^{th} **(C)** term of 4, 6, 8, ...) = [2 + (19)(2)][4 + (19)2] = (40)(42) = 1680**(B)** Given statement is not true for n = 1 as $1 < \left(\frac{1+1}{2}\right)^{r}$ i.e., 1 < 1 is not true. For n = 2, the statement becomes $1 \times 2 < \left(\frac{2+1}{2}\right)^2$ i.e., $2 < \frac{9}{4}$, which is true.

 \Rightarrow n(A \cup B) \geq 6

Since each question can be deal with in 3 ways

a) by selecting it $\Rightarrow \frac{\sin(\gamma+\delta) + \sin(\gamma-\delta)}{\sin(\gamma+\delta) - \sin(\gamma+\delta)}$ b) by selecting its alternative, or c) by rejecting it. $=\frac{\cos(\alpha-\beta)+\cos(\alpha+\beta)}{\cos(\alpha-\beta)-\cos(\alpha+\beta)}$ Thus, the total number of ways of dealing with 10 given questions is 3¹⁰ including a way in which we reject all the questions. (applying components and dividendo) Hence the number of all possible selections $\Rightarrow \frac{2\sin\gamma\cos\delta}{2\cos\gamma\sin\delta} = \frac{2\cos\alpha\cos\beta}{2\sin\alpha\sin\beta}$ of one or more questions is $3^{10} - 1$. We have $\tan \alpha + \tan \beta + \tan \gamma = 0$ 9. **(A)** $\Rightarrow \cot \delta = \cot \alpha \cot \beta \cot \gamma$ \Rightarrow tan $\gamma = -h$, and since tan γ satisfies the In case of R_1 , f(x) = 6x + 714. **(D)** given equation. Clearly, every element of A has a unique $\Rightarrow a \tan^{3} \gamma + (2a - x) \tan \gamma + y = 0$ image. \Rightarrow a(-h)³ + (2a - x)(-h) + v = 0 Hence, R_1 represents a function. \therefore ah³ + (2a - x)h = v Similarly, R_2 and R_3 also represent 10. **(A)** Let Nisha secure *x* marks in the fifth test, functions. then her average score In case of R_4 , $f(x) = \pm 4x$ $=\frac{87+92+94+95+x}{5}=\frac{368+x}{5}$: Every element of A has two unequal images. According to given condition, we must e.g., $f(1) = \pm 4$, $f(2) = \pm 8$ etc. have \therefore R₄ is not a function. $\Rightarrow \frac{368+x}{5} \ge 90$ $(1+4x)^{-5/4}.(1+2x)^{1/2}$ 15. **(D)** $=\left(1-\frac{5}{4}.4x\right)\left(1+\frac{1}{2}(2x)\right)$ \Rightarrow 368 + x > 90 \times 5 $\Rightarrow x > 450 - 368$ = (1 - 5x)(1 + x) $\Rightarrow x \ge 82$ \therefore Coefficient of x = 1 - 5 = -411. **(B)** $y = e^x$ and y = xNumber of triangles formed with 12 points 16. **(A)** $\Rightarrow e^x = x \Rightarrow no x \in \mathbf{R}$ $= {}^{12}C_3 = 220$ Number of triangles formed with 5 points Hence, $A \cap B = \phi$. $= {}^{5}C_{3} = 10$ 12. **(C)** Solving the given equations in pairs, we (Since, 5 points are collinear, so they get the coordinates of the vertices as cannot form a triangle) A = (-3, 3), B = (1, 1), C = (1, -1),Hence, total number of triangles D = (-2, -2)= 220 - 10 = 21017. **(B)** $n(A) = n(A - B) + n(A \cap B)$ $m_1 = \text{slope of AC} = \frac{-1-3}{1+3} = -\frac{4}{4} = -1$ = 30 + x + 2x= 30 + 3x $m_2 = slope of BD = \frac{-2-1}{2} = -\frac{-3}{2} = 1$ $n(B) = n(B - A) + n(A \cap B) = x + 2x = 3x$ n(A) = 2n(B) $m_1m_2 = (-1)(1) = -1.$ 30 + 3x = 2.3x \Rightarrow So, diagonals AC and BD are perpendicular. 30 = 6x - 3x \Rightarrow \therefore Angle between AC and BD is $\frac{\pi}{2}$. 30 = 3x \Rightarrow 13. **(A)** We have, $\cos(\alpha + \beta) \sin(\gamma + \beta) \sin(\gamma - \delta)$ x = 10 \Rightarrow

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 $\Rightarrow \frac{\sin(\gamma + \delta)}{\sin(\gamma - \delta)} = \frac{\cos(\alpha - \beta)}{\cos(\alpha + \beta)}$

18. (A)	We have,					
	$(1+x)^{n} = C_{0} + C_{1}x + C_{2}x^{2} + C_{3}x^{3} + C_{4}x^{4} + \dots + C_{n}x^{n} \dots (1)$					
	Differentiating w.r.t. x , we get					
	$n(1+x)^{n-1} = C_1 + 2C_2x + 3C_3x^2 + 4C_4x^3 + \dots + nC_xx^{n-1} \dots (2)$					
	Putting $x = 1$ in (1) and (2), we get					
	$2^{n} = C_{0} + C_{1} + C_{2} + C_{3} + C_{4} + \dots C_{n} \dots (3)$					
	and $n 2^{n-1} = C_1 + 2C_2 + 3C_3 + 4C_4 + + nC_n (4)$					
	Adding (3) and (4), we get					
	$(n+2)2^{n-1}{=}C_{_0}{+}2C_{_1}{+}3C_{_2}{+}{+}(n+1)C_{_n}.$					
19. (C)	Since, a, b and c are in A.P.					
	$\Rightarrow 2b = a + c$					
	Also, p is A.M. between a and b \dots (1)					
	\Rightarrow a, p, b are in A.P.					
	$\Rightarrow 2p = a + b$ (2)					
	Also, p'is G.M. between a and b					
	\Rightarrow a, p', b are in G.P.					
	$\Rightarrow p' = \sqrt{ab}$ (3)					
	Similarly,					
	2q = b + c (4)					
	and $q' = \sqrt{bc}$					
	$p^{2} = \frac{(a + b)^{2}}{4} = \frac{1}{4} (a^{2} + b^{2} + 2ab); p'^{2} = ab$					
	$q^{2} = \frac{(b+c)^{2}}{4} = \frac{1}{4} (b^{2} + c^{2} + 2bc); q'^{2} = bc$					
	$p^2 - q^2 = \frac{1}{4} (a^2 - c^2 - 2b(a - c)) = \frac{1}{4} (a^2 - c^2 + c^2)$					
	(a+c)(a-c)					
	$[\Box a, b, c \text{ are in A.P, } 2b = a + c]$					
	$=\frac{1}{2}(a^2-c^2)$					
	$p'^2 - q'^2 = ab - bc = b(a - c)$					
	$\Rightarrow \frac{(a+c)}{2} (a-c) = \frac{1}{2} (a^2 - c^2)$					
	:. $p^2 - q^2 = p'^2 - q'^2$					
20. (D)	The given expression					
$= \frac{\cos 6\theta + \cos 4\theta + 5\cos 4\theta + 5\cos 2\theta + 10\cos 2\theta + 10}{5\cos 2\theta + 10\cos 2\theta + 10\cos 2\theta + 10\cos 2\theta}$						
$\cos 5\theta + 5\cos 3\theta + 10\cos \theta$						

	$=\frac{2\cos^2}{2\cos^2}$	$5\theta\cos\theta + 5.2\cos3\theta\cos\theta + 10.2\cos\theta$	$\cos^2 \theta$					
		$\cos 2\theta + 2\cos 3\theta + 10\cos \theta$						
	$=\frac{2\cos\theta(\cos 5\theta+5\cos 3\theta+10\cos\theta)}{(\cos 5\theta+5\cos 3\theta+10\cos\theta)}=2\cos\theta$							
21.	(D)	Since α is a root of the equat	ion.					
		$a^2x^2 + bx + c = 0$						
		$\therefore a^2 \alpha^2 + b \alpha + c = 0$	(1)					
		Since β is a root of the equation	ion					
		$a^2x^2 - bx - c = 0$						
		$\therefore a^2\beta^2 - b\beta - c = 0$	(2)					
		Now let $f(x) = a^2x^2 + 2bx + 2c$						
		$\therefore f(\alpha) = a^2 \alpha^2 + 2(b \alpha + c)$						
		$=a^{2}\alpha^{2}+2(-a^{2}\alpha^{2})$	[from (1)]					
		$= -a^2 \alpha^2 < 0$						
	$f(\beta) = -a^2\beta^2 + 2(b\beta + c)$							
		$= a^2\beta^2 + 2(a^2\beta^2)$	[from (2)]					
		$= 3a^2\beta^2 > 0$						
		Since $f(\alpha)$ and $f(\beta)$ are of opposite sign and						
		γ is a root of the equation $f(x) = 0$,						
		$\therefore~\gamma$ must lie between α and	β.					
		Thus $\alpha < \gamma < \beta$.						
22.	(B)	The given digits are 0, 1, 2, 3, five in number.	4 which are					
		Since we are to form the numbers that are greater than 20000 and as no digits is to be repeated, every such numbers contains five digits and it must have 2, 3 or 4 at extreme left.						
		Thus, the extreme left place of up in 3 ways.	can be filled					
		Now the remaining four places up with the remaining four of ways. By using the principle of the required number of numb	can be filled digits in ⁴ P ₄ association, ers formed					
		$= 3 \times {}^{4}P_{4} = 3 \times \lfloor 4 = 3 \times 4 \times 3 \rfloor$	$\times 2 \times 1 = 72$					
23.	(C)	Statement P, 'Cardinality of a s is 1', is true.	singleton set					
		Statement Q. 'Cardinality of a	null set is 0'					
		is true.						

Hence (C) is the answer.

24. (C) Let h be the height of the triangle. Since, the area of the triangle is a^2

$$\begin{array}{l} \therefore \quad \frac{1}{2} \times a \times h = a^{2} \Rightarrow h = 2a \\ \text{Since the base lies along the line $x = a, \text{ the vertex lies on the line parallel to the base at a distance 2a from it. So, the required lines are $x = a \pm 2a$ i.e., $x = -a$ or $x = 3a \\ \text{25. (A)} \quad \text{The equation reduces to} \\ \qquad E = (-1)(-2)(-2)^{2} \dots (-2)^{15} \\ (x - 1)\left(x - \frac{1}{2}\right)\left(x - \frac{1}{2^{2}}\right) \cdots \left(x - \frac{1}{2^{15}}\right) \\ \Rightarrow E = 2^{1+2+3-3-.35} \\ \left(x - \frac{1}{2}\right)\left(x - \frac{1}{2^{2}}\right) \cdots \left(x - \frac{1}{2^{15}}\right) \\ \Rightarrow C = 2^{1120}\left\{\left(-1)\frac{1 - \left(\frac{1}{2}\right)^{15}}{1 - \left(\frac{1}{2}\right)^{15}}\right\} \\ = 2^{129}\left\{\left(-1\right)\frac{1 - \left(\frac{1}{2}\right)^{15}}{1 - \left(\frac{1}{2}\right)^{15}}\right\} \\ = -2^{105}\left\{\left(\frac{2^{16} - 1}{2 - 1}\right)\frac{2}{2^{16}}\right\} \\ = -2^{105}\left\{\left(\frac{2^{16} - 1}{2 - 1}\right)\frac{2}{2^{16}}\right\} \\ = -2^{105}\left\{\left(\frac{1 + i}{2}\right)^{2^{16}} = 1 \\ \Rightarrow \left(\frac{1 + i}{2}\right)^{2^{16}} = 1 \\ \Rightarrow \left(\frac{1 - 1 + 2i^{2}}{2}\right)^{2^{n}} = 1 \\ \Rightarrow n = 2 \text{ is the required smallest positive integer.} \\ 27. (C) \quad We have, f(\theta) = \sin \theta (\sin \theta + \sin 3\theta) \\ = \sin^{n} 2 (0 \ge 0, \text{ for all real } \theta \end{array}$$$$

v = 0TXGIN Let the last three numbers in A.P. be b, b + 6, b + 12 and the first number be a. Hence the four numbers are a, b, b + 6, b + 12 Given, a = b + 12.... (i) and a, b, b + 6 are in G.P. i.e., $b^2 = a(b + 6)$ \Rightarrow b² = (b + 12)(b + 6) [Since a = b + 12] or 18b = -72 $\therefore b = -4$ From (i), a = -4 + 12 = 8Hence the four numbers are 8, -4, 2 and 8. Since, $(a + ib)^5 = \alpha + i\beta$ (given) $\Rightarrow \left\{ \iota \left(\frac{a}{\iota} + b \right) \right\}^5 = \alpha + \iota \beta$ $\Rightarrow \iota^{5} (b - \iota a)^{5} = \alpha + \iota \beta \text{ (since } \iota = \frac{1}{\iota})$ $\Rightarrow \iota = (b - \iota a)^5 = \alpha + \iota \beta$ $\Rightarrow (b-\iota a)^{5} = \frac{\alpha+\iota\beta}{\iota} = \iota(\alpha+\iota\beta)$ \Rightarrow $(b-\iota a)^5 = \iota \alpha + \iota \beta$ Taking conjugate through out, we have $\left(\overline{b-\iota a}\right)^{5} = \left(\overline{\beta+\iota \alpha}\right)$ $\therefore (b + \iota a)^5 = \beta + \iota \alpha$

. (C) Let n be the number of newspapers which are read.

 $\Rightarrow 60n = (300)(5)$ $\therefore n = 25$

32. (A) Let orthocentre be
$$(\alpha, \beta)$$
.
Then, $3 = \frac{2(6)+1 \cdot \alpha}{2+1}$ and $3 = \frac{2(2)+1 \cdot \beta}{2+1}$
 $\Rightarrow \alpha = -3$ and $\beta = 5$
 \therefore Orthocentre is (-3, 5).
33. (B) We have,
 $\frac{\cos A}{\sin B \sin C} + \frac{\cos B}{\sin A \sin C} + \frac{\cos C}{\sin A \sin B}$
 $= \frac{\sin A \cos A + \sin B \cos B + \sin C \cos C}{\sin A \sin B \sin C}$
 $= \frac{\sin 2A + \sin 2B + \sin 2C}{2 \sin A \sin B \sin C}$
 $= \frac{4 \sin A \sin B \sin C}{2 \sin A \sin B \sin C} = 2$
34. (C) $A_1 = (2), A_2 = (2, 4), A_3 = (2, 4, 6)$ and so on.
Therefore $\prod_{n=1}^{207} A_n = (2)$
35. (A) $(^2C_0 + ^7C_1) + (^7C_1 + ^7C_2) + \dots + (^7C_6 + ^7C_7)$
 $= {}^8C_1 + {}^8C_2 + \dots + {}^8C_7$
 $= ({}^8C_0 + {}^8C_1 + {}^8C_2 + \dots + {}^8C_7) - ({}^8C_0 + {}^8C_8)$
 $= 2^8 - 1 (1 + 1) = 2^8 - 2$
36. (A) Let $x^{18} = y^{21} = z^{28} = k$. Then,
 $18 \log x = 21 \log y = 28 \log z = \log k$
 $\Rightarrow \log_y x = \frac{21}{18}, \log_x y = \frac{28}{21}, \log_x z = \frac{18}{28}$
 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 3 \log_z y = 4, 7 \log_x z = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x, 3 \log_2 y = 4, 7 \log_x z = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x, 3 \log_7 y = 4, 7 \log_x z = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x, 3 \log_7 y = 4, 7 \log_8 x = \frac{18}{28}$
 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{19}{2}$
 $\Rightarrow 3, 102, x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{18}{28}$
 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{18}{28}$
 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
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 $\Rightarrow 3, 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
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 $\Rightarrow 3, 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x = \frac{7}{2}, 3 \log_7 y = 4, 7 \log_8 x = \frac{9}{2}$
 $\Rightarrow 3, 3 \log_7 x = \frac{7}{2}, 3 \log_7 x = \frac{9}{2}, 0$
 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 3 \log_7 x = \frac{1}{2}, 0$
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 $\Rightarrow 3 \log_7 x = \frac{7}{2}, 0$
 $\Rightarrow 3 \log_7 x = \frac{7}{2},$

 $= p^{3} - 2(\alpha + \beta + \gamma)p^{2} + 4(\alpha \beta + \beta \gamma + \gamma \alpha)$ $p - 8\alpha\beta\gamma$ $= p^{3} + 4pq - 8r$ 39. **(C)** We have, sin(A + B + C) = 1, $\tan(A - B) = \frac{1}{\sqrt{3}}$ and $\sec(A + C) = 2$ \Rightarrow A+B+C=90°, A-B=30° and A+C=60° \Rightarrow B = 30°, A = 60° and C = 0° We have $\frac{1}{n(n+1)} = \frac{1}{n} - \frac{1}{n+1}$ 40. **(D)** $\therefore \quad \frac{1}{1\times 2} + \frac{1}{2\times 3} + \frac{1}{3\times 4} + \dots$ $+ \ \frac{1}{2007 \times 2008}$ $= 1 - \frac{1}{2} + \frac{1}{2} - \frac{1}{3} + \frac{1}{3} - \frac{1}{4} + \dots$ $+\frac{1}{2007}-\frac{1}{2008}$ $= 1 - \frac{1}{2008} = \frac{2007}{2008}$ **Physics** 41. **(B)** wood $u = 100 \text{ m s}^{-1}$ □ - □ m = 0.01 kg $v = 0 m s^{-1}$ $t = 0.02 \ s$ $u = 100 \text{ m s}^{-1}$ $v = 0 m s^{-1}$ a = ? $t = 0.02 \ s$ v = u + at $0 = 100 + a \times 0.02$ 0.02 a = -100a = -100 / 0.02 $a = -5000 \text{ m s}^{-2}$ A bullet penetrating a wooden block m = 0.01 kg $a = -5000 \text{ m s}^{-2}$ $\mathbf{F} = ?$ F = ma $F = 0.01 \, \times \, (- \, 5000)$ The average retarding force exerted by the wood is -50 N. 42. **(A)** $T_1 = 27 + 273 = 300 \text{ K}$

$$T_{2} = -13 + 273 = 260 \text{ K}$$
Coefficient of performance =
$$\frac{T_{2}}{T_{1} - T_{2}} = \frac{260}{300 - 260} = \frac{260}{40} = 6.5$$
43. (B) On the surface of the earth
$$g = \frac{GM}{R^{2}}; \text{ Weight mg} = 99 \text{ N}$$
At a height h above the earth
$$g' = \frac{GM}{(R+h)^{2}}, \text{ where } h = \frac{R}{2}$$

$$\frac{g'}{g} = \frac{R^{2}}{(R+h)^{2}} = \frac{R^{2}}{\left(R+\frac{R}{2}\right)^{2}} = \frac{R^{2}}{\frac{9}{4}} R^{2}$$

$$g' = \frac{4g}{9}$$
Weight = mg' = m $\times \frac{4g}{9} = \text{mg} \times \frac{4}{9}$
Here mg = 99 N = 99 $\times \frac{4}{9} = 44 \text{ N}$
44. (B) Mean diameter =
$$\frac{0.39 + 0.38 + 0.39 + 0.41 + 0.38 + 0.37 + 0.40 + 0.39}{8}$$

$$\overline{d} = 0.38875 \text{ mm}$$

$$= 0.39 \text{ mm} (rounded off to two significant figures)$$
Absolute error in the first reading =
$$0.39 - 0.39 = 0.00 \text{ mm}$$
Similarly finding the absolute error in the other seven readings and taking the mean;
Mean absolute error = $\overline{\Delta d} =$

$$\frac{0.00 + 0.01 + 0.00 + 0.02 + 0.01 + 0.02 + 0.01 + 0.00}{8}$$

$$= 0.00875 = 0.01 \text{ mm}$$
Relative error = $\frac{\overline{\Delta d}}{d} = \frac{0.01}{0.39} = 0.0256$
45. (B) Energy stored per unit volume
$$U = \frac{1}{2} \text{ stress} \times \frac{\text{strain}}{Y}$$

$$= \frac{1}{2} \text{ S} \times \frac{\text{S}}{Y} = \frac{1 \text{ S}^{2}}{2 \text{ Y}}$$
46. (B) A raw egg behaves like a spherical shell and a half boiled egg behaves like a solid sphere.

$$\therefore \frac{I_{r}}{I_{s}} = \frac{2/3 \text{ MR}^{2}}{2/5 \text{ MR}^{2}} = \frac{5}{3} > 1$$
47. (B) $x_{1} = 30 \times \frac{1}{2} \text{ km}$

$$x = x_{1} + x_{2} = 40 \text{ km}$$

$$v = \frac{x}{t} = \frac{40 \text{ km}}{1 \text{ h}} = 40 \text{ km h}^{-1}$$
(C) Mass m = 1500 kg, h = 50 m
t = 2×60 = 120 s
Power = $\frac{W}{t} = \frac{\text{mgh}}{t} = \frac{1500 \times 9.8 \times 50}{2 \times 60}$
= 6125 W
Power of the engine operating the lift

48.

$$= \frac{100 \times 6125}{75} = 8166.67 \text{ W}$$

49. (C) On the surface of the earth, the atmospheric pressure is quite high. The astronauts will feel great discomfort if they move on the earth immediately after coming back from the moon. To avoid it, they need to get used to normal air pressure gradually. That is why, they have to live for some days in a caravan with the air pressure lower than outside.

50. **(B)** Here
$$\frac{\Delta t}{t} = \frac{1}{10^{11}}$$

 $\Delta t = \frac{1}{10^{11}} \times t = \frac{1}{10^{11}} \times 10^{11} = 1$

or $\Delta t = 1 s$

Hence, maximum difference in time between two such clocks = 2 s

One may be 1 s faster and the other may be 1 s slower.

51. **(D)**
$$\frac{4 \text{ S}}{r_1} - \frac{4 \text{ S}}{r_2} = \frac{4 \text{ S}}{r}$$

or $\frac{1}{r} = \frac{1}{r_1} - \frac{1}{r_2} = \frac{1}{4} - \frac{1}{5} = \frac{1}{20}$ or $r = 20 \text{ cm}$

52. **(B)** As no external torque acts on the system, the angular momentum L is conserved. As the beads slide down, the moment of inertia of the system shall change. As $L = I \omega =$ constant and I changes, therefore, ω would change. As no work is being done, total energy cannot change.

53. (C)
$$a = \frac{dm}{dt} = -10 \text{ kg s}^{-1}$$
, $V_r = 5 \text{ km s}^{-1}$ (5000)
 $M = 1500 \text{ kg}$, $t = 50 \text{ s}$
 $\frac{10 \times 5000}{1500 - 10 \times 50} = 50 \text{ ms}^{-2}$
54. (A) Here, $m = 0.5 \text{ kg}$, $v = 1.5 \text{ m s}^{-1}$
 $K = 50 \text{ N m}^{-1}$
 $x = ?$
 $\frac{1}{2} \text{ K}x^2 = \frac{1}{2} \text{ mv}^2$
 $x = v\sqrt{\frac{m}{K}} = 1.5\sqrt{\frac{0.5}{50}} = 0.15 \text{ m}$
55. (A) Relative velocity of overtaking =

40 m s⁻¹ - 30 m s⁻¹ = 10 m s⁻¹.
Total distance covered with this relative
velocity during overtaking will be =
100 m + 200 m = 300 m.
Time taken t = 300 m/10 m s⁻¹ = 30 s
56. (D) Specific heat (in cal/g/°C): Copper (0.09)
Aluminium (0.21), Iron (0.11), Lead (0.03).
57. (B) Error in time period is

$$\Delta T = (0.1/20)s = 0.005 s$$
.
Also T = (20 s / 20) = 1s
Hence $\frac{\Delta T}{T} = \frac{0.005}{1} = 0.005 \times 100\% = 0.5\%$
58. (B) $C_m = \frac{3}{2}R, C_{di} = \frac{5}{2}R$.
If change in temperature is ΔT , then
 $1 \times \frac{3}{2} R \Delta T + 1 \times \frac{5}{2} R \Delta T = 2 \times C_v \times \Delta T$
This gives $C_v = 2 R$
59. (A) $g = \frac{GM}{R^2} = \frac{G}{R^2} \times \frac{4}{3}\pi R^3 \rho = \frac{4}{3}\pi GR \rho$,
i.e. $g \propto \rho$
 $\therefore \frac{g'}{g} = \frac{2\rho}{\rho} = 2$
or $g' = 2 g = 2 \times 9.8 = 19.6 m s^{-2}$
60. (C) $v = \frac{m_1 u_1 + m_2 u_2}{m_1 + m_2 u_2} = \frac{40 \times 4 + 60 \times 2}{40 + 60}$
 $= 2.8 m s^{-1}$
Loss in K.E. =
 $\frac{1}{2} m_1 u_1^2 + \frac{1}{2} m_2 u_2^2 - \frac{1}{2} (m_1 + m_2) v^2$
 $\frac{1}{2} [40 \times 16 + 60 \times 4 - 100 \times 2.8^2] = 48 J$
61. (A) Temperature of source =
 $T_1 = 100 + 273 = 373 K$
Temperature of sink =
 $T_2 = 30 + 273 = 303 K$
Efficiency = $\eta = \frac{T_1 - T_2}{T_1}$
 $= \frac{373 - 303}{373} = 0.188 = 18.8\%$
62. (A) The vertical displacement of the two stones
and their initial velocity acquired by them should
also be equal.
For the stone thrown vertically qownwards
 $a = -g, s = -h$
 $v_1^2 = u^2 + 2as$
 $v_1^2 = u^2 + 2(-g)(-h) = u^2 + 2gh$
For the stone thrown vertically downwards,

$$a = + g, s = h$$

$$v_2^2 = u^2 + 2gh$$

$$\frac{v_1^2}{v_2^2} = \frac{u^2 + 2gh}{u^2 + 2gh} = 1$$

$$v_1 : v_2 = 1 : 1$$
63. (C) Time taken in reaching bottom of incline is
$$t = \sqrt{\frac{2l(1 + K^2 / R^2)}{g \sin \theta}}$$
For solid cylinder (SC), K² = R² / 2
For hollow cylinder (HC), K² = R²
For solid sphere (S), K² = $\frac{2}{5}$ R²

$$\therefore t_s < t_{sc} < t_{Hc}$$
64. (B) Here $dx_1 = dx_2$, $A_1 = A_2$, $\frac{K_1}{K_2} = \frac{2}{3}$
Let θ be the temp. of the junction.
As
$$\frac{dQ_1}{dt} = \frac{dQ_2}{dt}$$

$$K_1 = (100 - \theta) = K_2 (\theta - 0)$$
or
$$\frac{K_1}{K_2} = \frac{\theta}{100 - \theta} = \frac{2}{3}$$

$$3\theta = 200 - 2\theta ; 5\theta = 200 ; \theta = 40 \text{ °C}$$
65. (C) m = 3 kg, u = 10 m s⁻¹, v = 0
Impulse = F × t = ?
Impulse = Change in momentum
$$F × t = m(v - u)$$

$$= 3 [0 - 10] = -30 \text{ N s}$$

$$\frac{Chemistry}{0}$$
66. (C) The set of elements given in options (A), (B)
and (D) have decreasing atomic radius.
Atomic radius in (Å)
Oxygen 0.73
Sulphur 1.09
Selenium 1.16
Tellurium 1.35
As the atomic number finceases within a
group, the atomic size increases
accordingly.
67. (D) All the alkali metals and their salts impart
colour to bunsen flame. The colours
imparted by different alkali metals are as
follows.
Element Li Na K Rb Cs
Colour Crimson Golden Pale Red Bluish
red yellow violet violet

When heat energy is supplied to alkali metal atom or ion in salt, the electronic excitation occurs in which electron jumps to higher energy level. When this excited electron deexcites to ground state, the energy is emitted in the form of electromagnetic radiation which lies in visible region thereby imparting colour to the flame. The colour of flame depends upon the wavelength of radiation emitted e.g., yellow D-line of Naspectra arises from $3s^1 \rightarrow 3p^1$ transition.

68. (A)
$$CO = N_2 = w g \therefore n_1 (CO) = \frac{w}{28}$$

 $n_2 (N_2) = \frac{w}{28}$. Hence, $P_{N_2} = P_{CO}$

*69. **(C)**

- 70. (**D**) Only coloured salts will form coloured metal metaborates.
- 71. **(B)** BF₃ is triangular planar and B_2H_6 is a dimer of triangular planar molecule (BH₃), therefore, both of these have zero dipole moment. NH₃ and NF₃, on the other hand have pyramidal structures and thus have dipole moments.



In NH₃, the dipole moments of the three N–H bonds reinforce the dipole moment due to lone pair of electrons but in NF₃, the dipole moments of the three N–F bonds oppose the dipole moment due to lone pair of electrons. As a result, dipole moment of NH₃ (μ = 1.46 D) is higher than that of NF₃ (μ = 0.24 D).

72. (B) Rise in temperature,

$$\begin{split} \Delta t &= (300.78 \ \mathrm{K} - 294.05 \ \mathrm{K}) = 6.73 \ \mathrm{K} \\ \mathrm{Heat \ capacity \ of \ the \ calorimeter =} \\ 8.93 \ \mathrm{kJ} \ \mathrm{K}^{-1} \\ \mathrm{Then,} \\ \mathrm{Heat \ transferred \ to \ calorimeter =} \\ \mathrm{Heat \ capacity \ of \ calorimeter \times Rise \ in \ temperature} \\ &= 8.93 \ \mathrm{kJ} \ \mathrm{K}^{-1} \times 6.73 \ \mathrm{K} \\ &= 60.1 \ \mathrm{kJ} \end{split}$$

- 73. (B) HNO₃ is added to decompose Na₂S and NaCN otherwise Na₂S will give black ppt. of Ag₂S and NaCN will give white ppt. of AgCN which would interfere with the test of halogens.
- 74. (A) 2-Ethylanthraquinol \rightarrow 2-Ethylanthraquinone + 2H₂O₂

75. (C)
$$K_{c} = \frac{[NO]^{2}}{[N_{2}O_{4}]} = \frac{(1.2 \times 10^{-2} \text{ mol } \text{L}^{-1})^{2}}{4.8 \times 10^{-2} \text{ mol } \text{L}^{-1}}$$

 $= 3 \times 10^{-3} \text{ mol } \text{L}^{-1}$

(A)
$$C: H = \frac{12 \times 100}{13 \times 12} = \frac{1 \times 100}{13 \times 1} = 1:1$$

 $\therefore E.F. = CH$

76

Since, P decolourises Br_2-H_2O , but Q does not, therefore, $P = C_2H_2$ (acetylene) and $Q = C_sH_s$ (benzene).

- 77. **(B)** According to Fajan's rule, the covalent bonding is maximum when W^+ is small and X^- is large.
- 78. (A) $\text{TiH}_{1.73}$ is a non-stoichiometric metallic or interstitial hydride.
- 79. (A) Except lime (50 60%), the major constituent of cement is silica (20 25%).
- 80. **(D)** (a) It is exact neutralisation. Hence, pH = 7.
 - (b) After neutralisation, $\frac{M}{10}$ HCl left = 10 ml. Total volume = 100 ml Dilution = 10 times. \therefore [H⁺] = 10⁻² or pH = 2 M
 - (c) After neutralisition, $\frac{M}{10}$ NaOH left = 80 ml. Total volume = 100 ml. pH > 7.
 - (d) After neutralisation, $\frac{M}{5}$ HCl left = 50 ml. Total volume = 100 ml Dilution = 2 times

*69. (C) The sum of mass % is 99.8. Hence, there is no oxygen in the given compound.

Element	Mass %	Atomic mass	Atomic ratio	Simplest ratio	Simplest whole number ratio
С	64.4	12	64.4 / 12 = 5.37	5.37 / 0.53 = 10.1	10
Н	5.5	1	5.5 / 1 = 5.5	5.5 / 0.53 = 10.4	10
Fe	29.9	56	29.9 / 56 = 0.53	0.53 / 0.53 = 1	1

Thus, the empirical formula of the compound is $C_{10}H_{10}Fe$.

$$\therefore [H^{+}] = \frac{1}{10} = 10^{-1} \text{ M or pH} = 1$$
81. (C) $BaO_2 + H_2SO_4 \rightarrow BaSO_4 + H_2O_2$
In this reaction, none of the elements undergo, a change in oxidation number or valency.
82. (A) Structure of B_2H_6 contains four 2c - 2e bonds and two 3c - 2e bonds.
83. (C) Molar mass of acetylene (C_2H_2),
 $M = (2 \times 12 + 2 \times 1)$ g/mol = 26 g/mol Mass of acetylene, m= 5.0 g
Temperature, T = $(50 \text{ °C} + 273) = 323 \text{ K}$
Pressure, P = 740 mm Hg = $\frac{740}{760}$ atm
 $= 0.9737$ atm
Using the gas equation,
 $PV = nRT = \frac{m}{M} RT$
 $V = \frac{m RT}{M P}$
 $\frac{5.0g \times 0.082 \text{ L atm K}^{-1} \text{mol}^{-1} \times 323 \text{ K}}{26 \text{ g mol}^{-1} \times 0.9737 \text{ atm}}$
 $= 5.23 \text{ L}$
84. (A) $c = v\lambda$ or $\lambda = \frac{c}{v} = \frac{3 \times 10^8 \text{ m s}^{-1}}{589 \times 10^{-9} \text{ m}}$
 $= 5.1 \times 10^{14} \text{ s}^{-1}$ (or Hz)

85. **(B)** % of S =
$$\frac{32}{233} \times \frac{0.233}{0.32} \times 100 = 10$$

86. (D) I has the tendency to lose as well as gain
electrons. Oxidation states of
$$Cs = +1$$

 $F = -1, 0$
 $Xe = Nil$
 $I = -1, 0, +1, +3, +5, +7$
87. (A) NH_4^+ is a conjugate acid of the base NH_3 .

- 88. **(B)** Cs with low IE is used in photoelectric cells.
- 89. (C) For PCl₃,

$$X = \frac{1}{2} = [VE + MA - c + a]$$
$$\frac{1}{2} [5 + 3 - 0 + 0] = 4$$
$$\therefore \text{ Hybridization of P in PC}l_3 \text{ is sp}^3.$$
For PCl₅.

$$X = \frac{1}{2} \left[5 + 5 - 0 + 0 \right] = 5$$

∴ Hybridization of P in PCl_5 is sp³d.

90. (B) Due to the poor shielding (screening) effect of d-electrons in case of Ga, the valence electrons are attracted more strongly and hence, the size is not increased.
